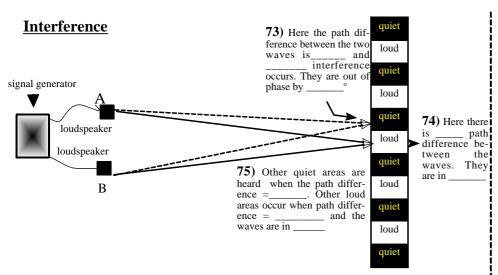
37) Kirchoff's second law states that in any closed loop the of the is equal to the e.m.f of the source. This law is a consequence of	37) sum p.d.'s energy
the conservation of	38) $R_1 + R_2 = 1/R_1 + 1/R_2$
38) For resistors in series $R_t = \dots$ In parallel $1/R_t = \dots$	
39) What is the resistance of this network	39) For parallel components
$ 10\Omega$ 10Ω	1/R = 1/10 + 1/10
	$1/R = 2/10 \text{ so } R = 5\Omega$
10Ω	total resistance = $10 + 5 = 15\Omega$
40) Use kirchoff's second law to calculate the current I in the circuit below	total resistance = $10 + 3 = 1322$
e.m.f = 10V	40) $10 = Ix2 + Ix2 + Ix2 = 6I$
2Ω Δ	so I = 10/6 = 1.67A
41) In a circuit with more than one cell e.m.f's are	41) Combined
42) In practise all sources of e.m.f have an resistance	42) Internal
43) The voltmeter reads the voltage, V across the load resistance and is called the p.d. It is less than the e.m.f. of the source because there is a potential difference across the of the source.	43) terminal internal resistance
From Kirchoff's second law $E = IR + \underline{\hspace{1cm}}$ $E = V + Ir$	Ir
If this is rearranged so $V = -Ir + E$	37 7
a graph of against would give a line with a gradient equal to the and the y intercept equal to the	V I straight internal resistance e.m.f
	44) potential divider
44) This is a circuit containing V _{in} a and a resistor.	thermistor variable resistor
$V_{out} = \underline{\hspace{1cm}}$. If the temperature increases the thermistor resistance $\underline{\hspace{1cm}}$ and the p.d.across R_1 will drop and the V_{out} will increase making the lamp	$V_{\text{out}} = \underline{V_{\text{in}}}\underline{R}_2$ drops brighter $(R_1 + R_2)$
45) In the potential divider circuit what is V_{out} if $V_{in} = 20V$, $R_1 = 500\Omega$ and $R_2 = 25\Omega$?	45) $V_{out} = V_{in}R_2 = 20x25/(500+25) = 0.95V$
46) What will happen to the lamp if R_2 is increased and	$(R_1 + R_2)$
why?	46) Gets brighter because V _{out} increases 47) second
47) Wave frequency is defined as the number of waves per	48) step step
48) Two wave are in phase if they are in with each other. Phase difference is a measure of how out of two waves could be. It can be described as a fraction of a eg $\lambda/2$ or as an angle 180° for example.	wavelength 49) wavelength amplitude
49) On this wave indicate the amplitude, wavelength and two points with zero displacement 50) The period, T of a wave is the time for	zero displacement 50) cycle or wave, T = 1/f
one and T = 51) The speed of a wave v =	51) $v = f\lambda$ λ – wavelength
52) Refraction of a wave occurs because of	
a change in when it enters a differ-	52) speed
ent medium. Diffraction is the out of waves through a Diffraction is greatest when the gap width is equal to	spreading gap wavelength

53) Two properties that all electromagnetic waves share is that they all travel at the speed of and they are all	53) light transverse 54) x ray(10 ⁻¹⁰ m), uv(10 ⁻⁸ m), visible (
54) The order of the electromagnetic spectrum in increasing wavelength is gamma (10 ⁻¹⁵ m)	5×10^{-7} m) infrared (10^{-5} m), microwave (10^{-2} m), radio(10^{-1+} m)
55) What is the frequency of gamma radiation with a wavelength of 10^{-15} m? $c = 3x10^8 \text{ms}^{-1}$	55) $f = c/\lambda = 3x10^8/10^{-15} = 3x10^{23}$ Hz
56) Electromagnetic waves with the largest have the most energy. Ultra-violet frequencies and above can atoms to make ions. This changes their chemistry and may lead to Infra-red, microwave and have a effect on objects. Microwaves of the right frequency can cause water molecules to resonate causing rapid heating.	56) frequency ionise cancers heating
57) UV part of the spectrum is split into UV-A,UV-B, UV-C, UV-A causes skin to, UV-B causes UV-C is mostly absorbed by in the atmosphere	57) wrinkle, cancer, ozone
58) Light is a wave of different In unpolarised light vibrations are in planes	58) transverse frequencies all
Polarised light is light where vibrations are in a single (typically vertical or). Light can be polarised using a	plane horizontal polariser
58b) Light emerging from this polarised	58b) vertically
polariser 59) When light reflects from a surface it is polarised. Sun-	59) partially
glasses contain a polariser which eliminates this polarised light.	
 60) The intensity of a wave is the power per metre. Its units are As an equation Intensity I = 61) The intensity of a wave is proportional to², I∝A² 	60) square Wm^{-2} , $\text{I} = \text{P/A}$ 61) amplitude
62) If the amplitude of a wave doubles its intensity If the amplitude of a wave halves its intensity reduces by	62) quadruples a quarter 63) $I_1/I_2 = A_1^2/A_2^2$
63) If a wave has amplitude A_1 and intensity I_1 and its amplitude is changed to A_2 then we can write $I_1/I_2 = \underline{\hspace{1cm}}$	$03/1_1/1_2 - 11_1/11_2$
second polariser (the analyser) second polariser (the analyser) rotated through 90°) no light emerges polariser wertically polarised through 90°) no light emerges Malus's law For rotations of the analyser between 0 and 90° the transmitted intensity I = I₀cos²θ where I₀ is the angle between the axis of the polariser and the plane of polarisation of the incident light	64) incident
65) The principal of superposition states that when two or more waves	65) the resulting displacement is the algebraic sum of the individual
These waves are in If combined there would be	displacements
interference	66) phase constructive
These waves are of phase by If combined there would be interference	67) out 180° destructive
68) For observable interference wave sources must be	68) coherent
69) Coherent waves are waves with the same	69) wavelength
70) The path difference between two waves is the difference in the two waves have travelled.	70) distance71) wavelengths interference
71) If the path difference between coherent waves is equal to a whole number of $\underline{\hspace{1cm}}$ (n λ) constructive $\underline{\hspace{1cm}}$ occurs	
72) If the path difference between two coherent waves is equal to $n\lambda/2$ interference occurs	72) destructive



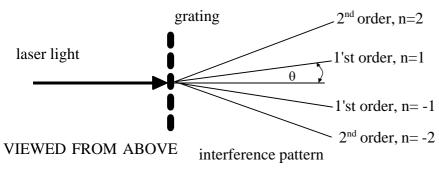
76) To observe interference of microwaves the same setup is used except a microwave ______ emits the signal. A barrier with two slits creates two _____ waves. These interfere and the interference pattern detected with a _____ detector.

77) In youngs slit experiment a laser is strikes a black slide with two parallel slits. This creates two ______ waves which interfere. The youngs slit equation is $\lambda =$ _____, where a is _____, x is _____ and D is _____

78) In youngs slit experiment if slit separation decreases fringe separation ______. If the slit screen distance increases the fringe separation ______

79) Youngs slit experiment is confirmation of the wave like nature of light because it is explained by the principle of ______ of waves.

80) A diffraction grating is a material with many slits. At each slit ______ occurs. Diffracted waves then interfere to produce an interference pattern



81) The diffraction equation is $d\sin\theta =$ _____. Where d is in metres, n is, and θ is the angle between the zeroth order (straight through) and the order number

82) If a diffraction grating has 3000 lines per cm the separation of each line = $\underline{\hspace{1cm}}$ cm = $\underline{\hspace{1cm}}$ m. If the first order maximum is detected at an angle of 10° as shown on the diagram what is the wavelength of the light?

83) Using a diffraction grating to measure the wavelength of light is better than youngs slit experiment because maxima are very _____ and _____. The maxima are widely separated and the angle can be measured very _____. In yougs slit experiment the separation of two slits needs to be measured - a big % inaccuracy. In a grating their are many slits and their separation d can be worked out accurately

73) $\lambda/2$ destructive 180

74) zero phase

75) $n\lambda/2$ $n\lambda$ phase

76) transmitter coherent microwave

77) coherent $\lambda = ax/D$ slit separation, fringe separation slit to screen distance

78) increases, increases

79) superposition

80) diffraction

81) $n\lambda$ slit separation, the order number

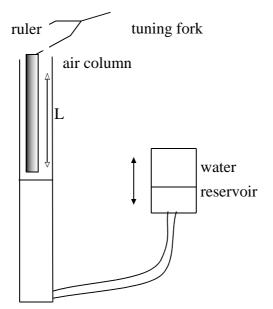
82)1/3000 = 3.33 x 10⁻⁴cm = 3.33x10⁻⁶m $\lambda = d\sin\theta/n = \frac{3.36x10^{-6}x\sin10}{1}$ = 5.8x10⁻⁷m

83) bright sharp accurately

84) White light shone through a diffraction grating will produce maxima of all the colours of the This happens because different colours have different and from dsinθ = nλ if the wavelength is bigger i.e. red light, the angle for maxima will be bigger	84) rainbow	wavelengths
85) White light is shone into this grating and an interference pattern observed. The zeroth order remains 86) Label the position of red R for the first order 87) Transverse and longitudinal are classed as waves. They transfer and seem to move Only can be polarised. Sound cannot be polarised because it is a wave.	85) white R 87) progressiv transverse	R e energy longitudinal
88) Two properties of stationary waves are (a) (b)	88) no energy they do not pro	
89) The vibrator sends waves down the string which with reflected waves. At certain frequencies large amplitudes of vibration are observed (resonance). Points of maximum displacement are called and zero displacement	89) interfere anti-nodes	nodes
The lowest frequency resonance occurs is called the The next frequency for resonance is call the harmonic 90) The distance between successive nodes is =	fundamental second 90) $\lambda/2$ 91) $\lambda/2$ 92) second $L = \lambda$	
quency is the harmonic. In terms of wavelength L = 93) In an open ended pipe the closed end is fixed and air cannot vibrate this is a If frequency is gradually increased a resonance occurs, (the fundamental). In terms of wavelength L = Increasing the frequency produces the second har-	93) node $L = \lambda/4$ $L = 3\lambda/4$	
monic and $L = \underline{\hspace{1cm}}$ node fundamental		

94a) To measure the speed of sound an experimenter changed the length of an air column by dropping or raising a water reservoir. The experimenter started with the air column very short then gradually increased it until maximum volume was heard

(a) sketch the stationary water wave obtained label the node and anti-node)



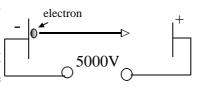
(b) use these results to work out the speed of sound

frequency $f = 512Hz$	Length $L = 0.16m$
-----------------------	--------------------

- 95) Electromagnetic radiation has a wave nature and a _____ nature.
- 96) Electromagnetic radiation travels as packets of energy called ______. Each photon has an energy E =_____ or E =_____
- 97) What is the energy of a gamma ray photon with wavelength 10⁻¹⁵m?
- 98) The electron volt is a much smaller unit of _____ than the joule
- 99) The electron volt is defined as the energy gained by an _____ when accelerated through a p.d. of 1 volt

from
$$W = QV$$
, $W = eV$, $= 1.6x10^{-19}x1 = 1.6x10^{-19}J$

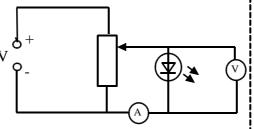
- 100) How many joules is 10eV?
- 101) How many eV's is $2x10^{-18}$ J?
- 102) Near the negative plate the electron has an electric potential energy, W. W = QV = eV. This electric potential energy is converted to _____ energy as it accelerates towards the positive plate. So $eV = \underline{mv}^2$



102b) What will be the velocity of the electron just before hitting if $V=5000V\,(m_e^{}=9.11x10^{-31}Kg)$

103) This apparatus can be used to calculate plancks constant,h.

The voltage is increased until a 6V _____ just starts to flow. At this threshold voltage



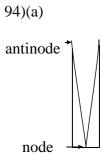
energy lost by electron = energy of emitted _____

so
$$eV = hf = hc/\lambda$$

so
$$V = \underline{hc}(1/\lambda)$$

e

If different diodes are used of known wavelength a graph of V against $1/\lambda$ will give a _____ line with a gradient = ____. Plancks constant h =



94b) $v = f\lambda$. From results at the fundamental $L = \lambda/4$

so
$$\lambda = 4L = 4x0.16 = 0.64m$$

so
$$v = f\lambda = 512x0.64 = 328ms^{-1}$$

- 95) particle
- 96) photons hf hc/λ

97) E =
$$hc/\lambda = (6.63 \times 10^{-34} \times 3 \times 10^{8})/10^{-15}$$

= $1.99 \times 10^{-11} J$

- 98) energy
- 99) electron

100)
$$10x1.6x10^{-19} = 1.6x10^{-18}J$$

101)
$$2x10^{-18}/1.6x10^{-19} = 12.5eV$$

102) kinetic

102b) v =
$$\sqrt{(2eV/m)}$$

= $\sqrt{(2x1.6x10^{-19}x5000/9.1x10^{-31})}$
= $4.2x10^7 ms^{-1}$

103) current

photon

straight hc/eh = (gradient x e)/c

104) In the photoelectric effect electrons are emitted from a metal surface using electromagnetic waves above a frequency. Below this frequency no electrons are emitted no matter how intense.	104) threshold
A wave model of electromagnetic waves cannot explain this because more intensity with a wave model implies more energy.	
If the intensity of electromagnetic waves above the threshold frequency is increased electrons are emitted but the maximum kinetic energy of the ejected electrons remains This is evidence of the particle nature of electromagnetic waves which we call the waves.	more the same
netic waves which we call with energy =	photons hf
105) The Einstein photo-electric equation is	105) $hf = \phi + 0.5 mv^2$
where hf is the, \phi is	photon energy work function
and 1/2mv ² is the maximum of an ejected electron	kinetic energy
106) At the threshold frequency $hf_0 = \underline{\hspace{1cm}}$	106) ¢
107) The work function is defined as	107) The minimum photon energy to just eject
108) (a) What are the energy of a photons in an electromagnetic wave with wavelength 2.4×10^{-7} m?	an electron 108) $E = hc/\lambda = (6.63 \times 10^{-34} \times 3 \times 10^{8})/2.4 \times 10^{-7}$
(b) If these electromagnetic waves are incident on the surface of a	$= 8.3 \times 10^{-19} \text{J}$
metal whose work function is 2.8x10 ⁻¹⁹ J calculate the maximum kinetic energy of electrons emitted.	(b) $0.5 \text{mv}^2 = \text{hf} - \phi = 8.3 \text{x} 10^{-19} - 2.8 \text{x} 10^{-19}$
(c) Determine the maximum speed of the emitted photons ($m_e = 9.11x10^{-31}Kg$)	$= 5.5 \times 10^{-19} \text{J}$ (c) $v = \sqrt{(2\text{KE/m})} = \sqrt{(2 \times 5.5 \times 10^{-19} / 9.1 \times 10^{-31})}$
109) <u>De Broglie</u> suggested if electromagnetic waves could behave like particles could particles behave as?	$= 1.1 \times 10^6 \text{ms}^{-1}$ 109) waves
110) Evidence that particles can behave like waves is the	110) diffraction
of electrons through a thin layer of graphite. 111) The De Broglie equation is	111) $\lambda = h/mv$
112) λ is a wave property, mass is a property (the product mv is momentum also a particle property)	112) particle
113) Calculate the de Broglie wavelength of an electron travelling at	113) $\lambda = h/mv = 6.63x^{-34}/(9.1x10^{-31}x10^7)$
10^7ms^{-1} . $(\text{m}_{\text{e}} = 9.11 \text{x} 10^{-31} \text{kg})$	$= 7.3 \times 10^{-11} \text{m}$
114) Will the electrons in the previous question be diffracted by solid materials (inter atomic spacing in solid materials is approx 10 ⁻¹⁰ m)	114) yes - wavelength and gaps similar
Emmision Spectra	
115) White light has a spectrum (all visible wavelengths are present). A hot gas like neon or sodium will produce a spectrum where only certain wavelengths are emitted.	115) continuous line
Absorption Spectra	
116) If white light shines through a gaseous element certain frequencies are absorbed and appear as lines against the continuous spectrum of the white light.	116) dark
117) When a photon is absorbed an electron is promoted to a higher	117) energy
level in the atom. When an electron falls back to a lower level it emits a	photon
118) The energy of the emitted photon is equal to the difference in level i.e. $hf = E_2 - E_1$	118) energy
level i.e. $hf = E_2 - E_1$ 119) If the electron at n=1, the ground state, absorbed a photon of energy 47ev it would be 120) What would the energy of an red reduction at $n=3$ $n=4$ n	119) just removed from the atom
120) What would the energy of an emitted photon if it fell from $n = 3$ to $n = 2$?	120) hf = E_2 - E_1 = -13-(-22) = 9eV
121) What would the frequency be of the emitted photon?	121) E=hf, $f = E/h = 9x1.6x10^{-19}/6.63x^{-34}$ = $2.2x10^{15}$ Hz